ALKYLATION OF SCHLEYER'S LITHIOCARBON C<sub>4</sub>Li<sub>4</sub> BY METHYL IODIDE
Nikolai S.Zefirov, V.N.Kirin, Nina M.Yur'eva, A.S.Koz'min, N.S.Kulikov and Yu.N.Luzikov
Department of Chemistry, Moscow State University, Moscow 117234 USSR

Summary - Alkylation of lithiocarbon C4 Li4 gives the hydrocarbon C8 H12 to which the structure of tetramethyltetrahedrane has been ascribed.

The problem of tetrahedrane has fundamental importance from the theoretical point of view. Inspite of fairly extensive literature (see reviews 1) the attempts at synthesis of this unique structure remains unsuccessful until recent times. In 1978 there appeared two preliminary communications claiming success with this long pursued goal 2,3. G.Maier and co-workers 2 described the synthesis of tetra-tert-butyltetrahedrane starting from the corresponding homotetrahedrane-ne. P.von R.Schleyer and co-workers described the synthesis of tetralithium derivative of tetra-hedrane (1, C4Li4) by photolysis of LiC CLi 3. The idea for that synthesis came as the result of the extensive theoretical investigation of polylithiocarbons and related compounds by the group of P.von R.Schleyer and J.Pople 4.

In the course of our study of cage compounds <sup>5</sup>, it occured to us that compound 1 could, in principle, be used as a starting material for the synthesis of tetrahedrane derivatives. To prove this idea, we investigated the alkylation of 1 by methyl iodide, using the following procedure:

10 ml of CH<sub>3</sub>I was added dropwise (20 min) to a stirred suspension of solid 1 (prepared from 0.25 mol of Li in accordance with Ref.3) in 100 ml of NH<sub>3</sub>. The mixture was kept at room temperature to evaporate the NH<sub>3</sub>, diluted with 250 ml of water and extracted with pentane. After the careful evaporation of pentane, 0.2-0.5g of colourless residue was obtained, representing the mixture of pentane and new compound (2:1).

The IR spectrum of the mixture contained the bands 1380,1465,2863 and 2935 cm $^{-1}$ . The <sup>1</sup>H NMR spectrum of the mixture showed a new signal at  $\delta$  1.26 ppm, which is absent in a pure pentane.

The <sup>13</sup>C NMR spectra of the mixture showed two new peaks at 29.51 ppm(quaternary carbon atom) and at 29.46 ppm (CH<sub>3</sub> group); the assignment has been made by off-resonance. The mass-spectra (both electron and field ionization) of this mixture has shown peaks with m/e 108(M), 107(M-1) and 92(M-CH<sub>3</sub>-H). These data forced us to conclude that alkylation of 1 by methyl iodide leads to the formation of tetramethyltetrahedrane 2. This work demonstrate, in principle, a promising synthetic way to obtain the tetrahedrane structure via tetralithium derivative 1; this investigation is now actively underway.

$$\begin{array}{c|c}
\text{Li} & & & \\
\text{C} & & \\
\text{C} & & \\
\text{Li} & & \\
\end{array}$$

$$\begin{array}{c|c}
\text{CH}_3 \\
\text{CH}_3 \\
\text{CH}_3 \\
\text{CH}_3 \\
\text{CH}_3
\end{array}$$

$$\begin{array}{c}
\text{CH}_3 \\
\text{CH}_3 \\
\text{CH}_3 \\
\text{CH}_3
\end{array}$$

References and notes

- N.S.Zefirov, A.S.Koz<sup>\*</sup>min, A.V.Abramenkov, <u>Usp.Khim(Chem.Rev.Russ.</u>) 47,289(1978), Engl.Transl., 47,163(1978); J.F.Libman, A.Greenberg, <u>Chem.Rev.</u>, 76,311(1976).
- G.Maier, S.Pfriem, U.Schafer and R.Matusch, Angew. Chem. Int. Ed. Engl., 17,520(1978).
- G.Rauscher, T. Ceark, D. Poppinger and P. von R. Schleyer, Angew. Chem., 90,306(1978).
- 4 E.S.Jemmis, D.Poppinger, P.von R.Schleyer and J.A.Pople, J.Am.Chem.Soc., 99,5796(1977); J.D.Dill P.von R.Schleyer, J.S.Binkley and J.A.Pople, <u>Ibid</u>., 99,6159(1977).
- N.V.Averina, A.M.Boganov, S.N.Vlasova, N.S.Zefirov, A.S.Koz'min and S.S.Yarovoi, Zh.Org.Khim., 12, 2124(1976); N.S.Zefirov, N.V.Averina and A.M.Boganov, Ibid., 14,966(1978); N.S.Zefirov, N.V. Averina, A.M.Boganov, T.S.Kuznetsova and S.S.Yarovoi, Dokl.Akad.Nauk SSSR, 233,373(1977).
- Electron diffraction experiments as well as attempts to obtain low temperature X-ray diffraction data are in progress.

(Received in UK 7 March 1979)